

## **New Tetrasaccharide Flavonol Glycoside from *Epimedium acuminatum***

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*J. Nat. Prod.*, **1992**, 55 (5), 672-675 • DOI:

10.1021/np50083a019 • Publication Date (Web): 01 July 2004

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NEW TETRASACCHARIDE FLAVONOL GLYCOSIDE FROM  
*EPIMEDIUM ACUMINATUM*

BI-HUANG HU,\* LI-DONG ZHOU, and YONG-LONG LIU

*Institute of Medicinal Plant Development, Chinese Academy of Medical Sciences,  
Beijing 100094, People's Republic of China*

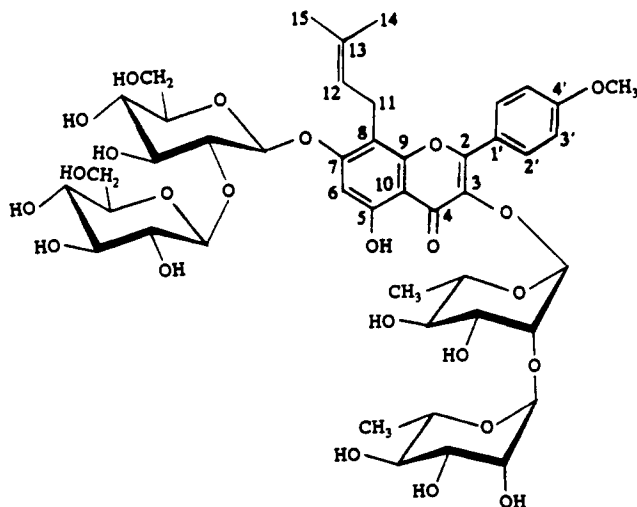
**ABSTRACT.**—A new tetrasaccharide flavonol glycoside was isolated from the aerial parts of *Epimedium acuminatum*, along with three known flavonoids. The structure of the new compound, named acuminatoside [1], was established to be anhydroicaritin-3-O- $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 2)- $\alpha$ -L-rhamnopyranoside-7-O- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 2)- $\beta$ -D-glucopyranoside by means of spectroscopic techniques (uv, eims, fdms, fabms,  $^1\text{H}$  nmr,  $^1\text{H}$ - $^1\text{H}$  COSY, 2D-J,  $^{13}\text{C}$  nmr, APT, and  $^1\text{H}$ - $^{13}\text{C}$  HETCOR) and chemical methods (acid hydrolysis, enzymatic hydrolysis, and tlc-densitometry). The known compounds were identified as icariin, epimedeside A, and kaempferitrin.

Some plants of the genus *Epimedium* (Berberidaceae) have been used as a tonic in traditional Chinese medicine. In a previous study, we investigated the constituents of the  $\text{CHCl}_3$ -soluble and EtOAc-soluble portions of the 95% EtOH extract from the aerial part of *Epimedium acuminatum* Franch (1). We report here the structure elucidation of a new flavonoid, acuminatoside [1], and the identification of three known flavonoids, isolated from the *n*-BuOH-soluble portion of the above extract.

## RESULTS AND DISCUSSION

Acuminatoside [1], a yellow amor-

phous powder, was positive to Mg-HCl and Molish tests. The uv spectrum showed absorption bands at 266, 311, and 338 nm, and bathochromic shifts were observed after adding certain reagents (NaOMe,  $\text{AlCl}_3/\text{HCl}$ , and NaOAc), which indicated that 1 was a flavonoid with the presence of a 5-hydroxyl group and the absence of free hydroxyl groups at C-3, C-4', and C-7. The eims gave the molecular ion peak of the aglycone of 1 at  $m/z$  368. The fragments ( $m/z$  353, 313, 165, and 135) formed after retro-Diels-Alder cleavage suggested that 1 had a prenyl group in ring A and an MeO group in ring B.



Fabms data showed that **1** contained four sugar units. In the  $^1\text{H}$ -nmr spectrum (300 MHz,  $\text{DMSO-}d_6$ ), four anomeric protons were observed and assigned to those of rhamnose  $\delta$  5.37 (br, s), glucose 5.00 (d,  $J=8.0$  Hz), rhamnose 4.86 (br, s), and glucose 4.28 (d,  $J=8.0$  Hz). The presence of two two-proton doublets at  $\delta$  7.87 ( $J=7.5$  Hz) and 7.11 ( $J=7.5$  Hz) and a one-proton singlet at 6.61, assignable to the proton at C-6 because of a glucosylation at C-7 (2), suggested that the aglycone was based on kaempferol with a substituent carbon linked at C-8. The characteristic signals based on a prenyl group as the substituent were observed at  $\delta$  1.66 (3H, s), 1.58 (3H, s), 3.34 (2H, m), and 5.14 (1H, br, t,  $J=5.0$  Hz). The location of the prenyl group at the C-8 position was supported by the  $^{13}\text{C}$ -nmr spectrum of **1** because of the chemical shift value of the carbon atom at the C-6 position ( $\delta$  98.3) (3). In addition to those, the signal of an MeO group was observed at  $\delta$  3.83 (3H, s). Combined with the eims fragments, the aglycone was concluded to be anhydroicaritin.

Acid hydrolysis of **1** afforded D-glucose, L-rhamnose, and the aglycone. The molar ratio of aglycone to glucose to rhamnose from **1** was 1:2:2, which was determined by tlc-densitometry after acid hydrolysis. Partial acid hydrolysis of **1** produced a compound which was identified as anhydroicaritin-7-O- $\beta$ -D-glucopyranoside by spectroscopic analysis.

Enzymatic hydrolysis of **1** gave a product. The uv spectrum of this product and the shift values after adding some reagents showed that the hydroxyl group at C-7 was free. The fabms revealed that the product contained two rhamnose units. Combined with the uv and  $^1\text{H}$ -nmr results, the disaccharide moiety was attached at C-3. The interlinkage of the biose was determined to be  $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 2)- $\alpha$ -L-rhamnopyranoside by a key  $^{13}\text{C}$ -nmr signal (75.5 ppm) and was confirmed by

the  $^1\text{H}$ - $^{13}\text{C}$  HETCOR spectrum which showed that the proton ( $\delta$  4.14) at the C-2 of the rhamnose attached to the aglycone had a cross peak with the C-2 ( $\delta$  75.5), indicating that the carbon signal was shifted downfield  $\delta$  5.4 by the glycosylation of the terminal rhamnose. The enzymatic hydrolysis product was, therefore, identified as anhydroicaritin 3-O- $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 2)- $\alpha$ -L-rhamnopyranoside.

So far, the above data revealed that **1** contained the structure of anhydroicaritin-3-O- $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 2)- $\alpha$ -L-rhamnopyranoside-7-O- $\beta$ -D-glucopyranoside. Further, the  $^{13}\text{C}$ -nmr spectrum of **1** showed four anomeric carbon signals of the sugar moieties at  $\delta$  101.8, 100.8, 100.7, and 97.0. The location of the other glucose unit of **1** was determined to be attached to the glucose unit at the C-7 position by its  $^{13}\text{C}$ -nmr and  $^1\text{H}$ - $^{13}\text{C}$  HETCOR spectra. The chemical shift value at the C-2 ( $\delta$  82.0) of the glucose unit attached to aglycone was shifted downfield ( $\delta$  8.5), but those at the C-1 ( $\delta$  97.0) and C-3 ( $\delta$  76.6) showed upfield shifts (ca.  $\delta$  2.4 and 0.2, respectively), demonstrating that the terminal glucose was linked to the glucose unit by a (1 $\rightarrow$ 2) linkage (4).

From these data, the structure of **1** was concluded to be anhydroicaritin-3-O- $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 2)- $\alpha$ -L-rhamnopyranoside-7-O- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 2)- $\beta$ -D-glucopyranoside, and was named acuminatoside. In addition, the three known compounds were identified as icariin, epimedeside A, and kaempferitrin on the basis of spectroscopic analysis and comparison with authentic samples.

## EXPERIMENTAL

GENERAL EXPERIMENTAL METHODS.—The  $^1\text{H}$ -nmr and  $^{13}\text{C}$ -nmr spectra measured with TMS as internal reference were run on FX-100, XL-VXR 300, and JNM-GX 400 nmr spectrometers, respectively. Eims and fdms were measured on a MAT 711 mass spectrometer. Fabms were recorded on a KYKY ZhP-5 mass spectrometer. Uv spectra were measured on a Philips PYS Unicam PU 8800. Sephadex LH-20

(Pharmacia), Polyclar A T (Serva), and cellulase (Sigma) were used. Polyamide films were produced by the Huang-Yan Chemical Factory, Zhejiang Province, China. Si gel was the product of the Qingdao Marine Chemical Factory, Shandong Province, China.

**PLANT MATERIAL.**—The aerial parts of *E. acuminatum* were collected on the E-mei Mountain, Sichuan Province, China in July, 1985. The material was identified by Prof. Wen-yan Lian, Institute of Medicinal Plant Development, Chinese Academy of Medical Sciences. A voucher specimen has been deposited in the herbarium of the Institute.

**EXTRACTION AND ISOLATION.**—The dried aerial parts (2 kg) were extracted 4 × with 95% EtOH (24 liters) under reflux for 2 h. The combined extracts were concentrated at reduced pressure. The residue (292 g) was suspended in H<sub>2</sub>O and extracted successively with CHCl<sub>3</sub>, EtOAc, and *n*-BuOH. The *n*-BuOH portion (49 g) was chromatographed on a Si gel Column (eluent: CHCl<sub>3</sub>-MeOH gradient) and separated into seven parts. Compound **1** (1.5 g) from part five, icariin (2.0 g) from part three, and epimedeside A (100 mg) and kaempferitrin (50 mg) from part four were obtained by the combination of Polyclar AT CC and purification with a Sephadex LH-20 column.

TABLE 1. <sup>13</sup>C-nmr Spectra<sup>a</sup> of Compound **1** and Its Enzymatic Hydrolysis Product.

Carbon	Compound			
	<b>1</b>		Enzymatic hydrolysis product of <b>1</b>	
C-2	157.3		156.7	
C-3	134.6		134.4	
C-4	178.2		177.9	
C-5	160.6		161.9	
C-6	98.3		98.4	
C-7	161.5		161.3	
C-8	108.4		106.0	
C-9	153.0		153.8	
C-10	105.7		104.2	
C-11	21.7		21.2	
C-12	122.2		122.3	
C-13	131.2		131.3	
C-14	25.7		25.4	
C-15	18.1		17.8	
C-1'	122.2		122.3	
C-2', -6'	130.7		130.4	
C-3', -5'	114.2		114.1	
C-4'	159.1		158.9	
OMe	55.7		55.5	
Rha				
C-1	100.7	101.8	100.7	101.6
C-2	75.7	70.3	75.5	70.1
C-3	70.4	70.6	70.5	70.7
C-4	72.1	71.5	72.0	71.4
C-5	69.8	69.0	70.3	68.8
C-6	17.9	17.8	17.6	17.5
Glu				
C-1	97.0	100.8		
C-2	82.0	73.5		
C-3	76.7	76.9		
C-4	70.0	69.4		
C-5	77.3	77.3		
C-6	61.4	60.8		

<sup>a</sup>Chemical shifts were expressed in ppm from TMS. Solvent DMSO-*d*<sub>6</sub>; **1** (75.45 MHz), enzymatic hydrolysis product (100.6 MHz). Assigned on the basis of <sup>1</sup>H-<sup>1</sup>H COSY and <sup>1</sup>H-<sup>13</sup>C HETCOR spectra.

IDENTIFICATION OF KNOWN FLAVONOIDS.—Icariin: yellow needles; mp 245–247°; Mg-HCl test positive; Molish test positive. Icariin was identified by uv, eims,  $^1\text{H}$  nmr, and direct comparison with an authentic specimen (ir and tlc). Epimedeside A: yellow amorphous powder; mp 219–221°; Mg-HCl test positive; Molish test positive. Epimedeside A was identified by uv, fdms, eims,  $^1\text{H}$  nmr, and  $^{13}\text{C}$  nmr. Kaempferitrin: yellow needles; mp 196–196.5°; Mg-HCl test positive; Molish test positive. Kaempferitrin was identified by uv, fdms, eims,  $^1\text{H}$  nmr, and  $^{13}\text{C}$  nmr.

ACID HYDROLYSIS OF **1**.—A 10% HCl solution (5 ml) of **1** (25 mg) was heated under reflux for 10 h. The solution was neutralized with  $\text{Ag}_2\text{CO}_3$ , and the filtrate was subjected to Si gel tlc [developing solvent  $\text{CHCl}_3\text{-MeOH-H}_2\text{O}$  (13:7:2) lower phase], to result in identification of glucose and rhamnose by the usual procedures.

To determine the molar ratio of the sugars and the aglycone of **1**, a 10% HCl solution of **1** was heated under reflux for 10 h. The solution was evaporated in vacuum, and the residue was prepared for quantitative determination by tlc densitometry (5). For partial acid hydrolysis of **1** a 15% HOAc solution (10 ml) of **1** (100 mg) was heated under reflux for 10 h. The precipitate was purified by a Sephadex LH-20 column and recrystallized to produce the partial hydrolysis product.

ENZYMATIC HYDROLYSIS OF **1**.—A 0.2 M NaOAc/HOAc buffer solution (pH 5) (50 ml) that contained **1** (300 mg) and cellulase (300 mg) was incubated at room temperature for 48 h. The precipitate was chromatographed on a polyamide column and purified with Sephadex LH-20, giving a yellow powder of the enzymatic hydrolysis product.

ACUMINATOSIDE [**1**].—A yellow amorphous powder: uv  $\lambda$  max (MeOH) nm 266, 311, 338 sh, (+ NaOMe) 268, 294 sh, 350, (+  $\text{AlCl}_3$ ) 276, 303, 340, 405, (+  $\text{AlCl}_3/\text{HCl}$ ) 274, 301, 336, 404, (+ NaOAc) 268, 312, 336 sh, (+ NaOAc/ $\text{H}_3\text{BO}_3$ ) 266, 312, 336 sh; fdms  $m/z$  [ $\text{M} + \text{Na} + 2\text{H}$ ] $^+$  1009, [ $\text{M} - \text{glu} + \text{Na} + 2\text{H}$ ] $^+$  847, [ $\text{M} - 2 \text{glu} + \text{Na} + 2\text{H}$ ] $^+$  685, [aglycone + glu + H] $^+$  531, [aglycone] $^+$  368; eims  $m/z$  368 (100), 353 (32.5), 339 (2.2), 325 (2.0), 313 (74.2), 300 (5.6), 284 (2.0), 184 (7.2), 165 (7.3), 157 (17.0), 135 (10.7), 128 (6.9), 107 (5.3);  $^1\text{H}$  nmr (300 MHz,  $\text{DMSO}-d_6$ )  $\delta$  0.79 (3H, d,  $J = 6.0$  Hz, rha-Me'), 1.08 (3H, d,  $J = 6.0$  Hz, rha-Me), 1.58, 1.66 (6H, s x2, Me-14, -15), 3.34 (2H, m, H-11), 3.83 (3H, s, 4'-OMe), 4.10 (1H, br, rha-H-2), 4.28 (1H, d,  $J = 8.0$ , glu-H-1'), 4.86 (1H, br, s, rha-H-1'),

5.00 (1H, d,  $J = 8.0$  Hz, glu-H-1), 5.14 (1H, br, t,  $J = 5.0$  Hz, H-12), 5.37 (1H, br, s, rha-H-1), 6.61 (1H, s, H-6), 7.11 (2H, d,  $J = 7.5$  Hz, H-3'; 5'), 7.87 (2H, d,  $J = 7.5$  Hz, H-2'; 6'), 12.60 (1H, s, 5-OH);  $^{13}\text{C}$  nmr see Table 1.

PARTIAL ACID HYDROLYSIS PRODUCT OF **1**.—A yellow powder: uv  $\lambda$  max (MeOH) nm 266, 320, 369, 420 sh, (+ NaOMe) 261, 419, (+  $\text{AlCl}_3$ ) 264, 296 sh, 354, 432, (+  $\text{AlCl}_3/\text{HCl}$ ) 262, 294 sh, 351, 430, (+ NaOAc) 266, 321, 372, (+ NaOAc/ $\text{H}_3\text{BO}_3$ ) 267, 321, 369; fabms  $m/z$  [aglycone + glu + 1] $^+$  531, [aglycone + 1] $^+$  369.

ENZYMATIC HYDROLYSIS PRODUCT OF **1**.—A yellow powder: uv  $\lambda$  max (MeOH) nm 266, 284 sh, 334 sh, (+ NaOMe) 278, 376, (+  $\text{AlCl}_3$ ) 275, 304, 342, 402 sh, (+  $\text{AlCl}_3/\text{HCl}$ ) 276, 300, 338, 400 sh, (+ NaOAc) 276, 352, (+ NaOAc/ $\text{H}_3\text{BO}_3$ ) 266, 284 sh, 332 sh; fabms  $m/z$  [aglycone + 2 rha + 1] $^+$  661, [aglycone + rha + 1] $^+$  515, [aglycone + 1] $^+$  369,  $^1\text{H}$  nmr (400 MHz,  $\text{DMSO}-d_6$ )  $\delta$  0.83, (3H, d,  $J = 6.0$  Hz, rha-Me'), 1.13 (3H, d,  $J = 6.0$  Hz, rha-Me), 1.64, 1.69 (6H, s x2, Me-14, -15), 3.32, 3.44 (each 1H, br, d,  $J = 14.0$  Hz, H-11), 3.71 (1H, br, s, rha-H-2'), 3.87 (3H, s, 4'-OMe), 4.14 (1H, br, s, rha-H-2), 4.91 (1H, br, s, rha-H-1'), 5.17 (1H, br, t,  $J = 5.0$  Hz, H-12), 5.39 (1H, br, s, rha-H-1), 6.34 (1H, s, H-6), 7.13 (2H, d,  $J = 9.0$  Hz, H-3', -5'), 7.87 (2H, d,  $J = 9.0$  Hz, H-2', -6');  $^{13}\text{C}$  nmr see Table 1.

#### ACKNOWLEDGMENTS

The authors are grateful to The National Foundation of Natural Sciences, China, for supporting this project. We also thank Prof. Dr. Xiao-tian Liang, Institute of Materia Medica, Chinese Academy of Medical Sciences, for helpful advice.

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Received 15 February 1991